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Methods for determination of pH of aqueous solutions

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Foreword

This translation has been made based on the original Japanese Industrial Standard revised by the Minister of Economy, Trade and Industry through deliberations at the Japanese Industrial Standards Committee as the result of proposal for revision of Japanese Industrial Standard submitted by the Society of Instrument and Control Engineers (SICE)/Japanese Standards Association (JSA) with the draft being attached, based on the provision of Article 12 Clause 1 of the Industrial Standardization Law applicable to the case of revision by the provision of Article 14.

Consequently JIS Z 8802:1984 is replaced with this Standard.

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Methods for determination of pH of aqueous solutions

JIS Z 8802:2011

1 Scope

This Japanese Industrial Standard specifies the methods for determination of pH of aqueous solutions of 0 °C to 95 °C with the pH meter using glass electrode.

2 Normative references

The following standards contain provisions which, through reference in this text, constitute provisions of this Standard. The most recent editions of the standards (including amendments) indicated below shall be applied.

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JIS K 0050 General rules for chemical analysis

JIS K 0211 Technical terms for analytical chemistry (General part)

JIS K 0213 Technical terms for analytical chemistry (Electrochemistry part)

JIS K 8474 Potassium trihydrogen dioxalate dihydrate (Reagent)
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JIS K 8514 Sodium bromide (Reagent)

JIS K 8622 Sodium hydrogen carbonate (Reagent)

JIS K 8625 Sodium carbonate (Reagent)

JIS K 8809 Potassium hydrogen phthalate (Reagent)

JIS K 8866 Sodium tetraborate decahydrate (Reagent)

JIS K 9007 Potassium dihydrogen phosphate (Reagent)

JIS K 9020 Disodium hydrogenphosphate (Reagent)

JIS Z 8805 Glass electrodes for measurement of pH

3 Terms and definitions

For the purposes of this Standard, the terms and definitions given in **JIS K 0211**, **JIS K 0213** and **JIS Z 8805**, and the following apply.

3.1 pH

the value obtained from the electromotive force measured with a glass electrode pH meter utilizing the pH value of pH standard solution specified in this Standard as reference

NOTE: It is read as PÎETTI or PÎEITI.

3.2 certified pH standard solution

the pH standard solution of which the pH value is measured according to the primary method determined by CCQM of CIPM, or the pH standard solution being traceable to it in which the expanded uncertainty to give $95\,\%$ of confidence interval is approximately within 0.015